206. Experiments on the Total Synthesis of Lysolipin I

Part II¹)

Michael Addition of 1,3-Cyclohexanedione to Quinone Acetals²)

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Summary

Base-catalyzed reaction of 1,3-cyclohexanedione (3) with the quinone monoacetals 4 and 7 leads to the polycyclic products 5 and 8, respectively, and in the case of 4 to variable amounts of dibenzofuranone 6. The 2-arylcyclohexanedione 9, on the other hand, is isolated from the reaction of 3 and bisacetal 11 catalyzed by $ZnCl_2$ (Scheme 2). Treatment of the adduct 8 with $(CH_3O)_2SO_2/K_2CO_3$ results in cleavage of the heterocyclic ring by a *retro-Michael* reaction affording the labile enone 23 which was further transformed to 24 by selective hydrogenation. The 8-acetoxydibenzofuranone 22 is obtainable from 8 by acid treatment and acetylation (Scheme 4). The reactions of the silylenol ethers 27 and 35 with quinone monoacetals were very complex (Scheme 6). The desired arylcyclohexanone derivatives 28 and 36 were formed in very low yields. Under certain conditions (elevated temperature or strong Lewis acids as catalysts), single-electron transfer or addition to the ene-acetal rather than to the enone function of the quinone monoacetals became predominant. In connection with this study, the sensitive 2-methoxy-p-benzoquinone monoacetals 15 (Scheme 3) and 29 (Scheme 6) have been prepared and characterized.

1. Introduction. – 2'-Aryl-3'-oxocyclohexylacetic acids of structure **a** are key compounds in our approach [1] to the total synthesis of Lysolipin I (1) [2] (Scheme 1). Such 2-arylcyclohexanones have generally been prepared by the addition of an aryllithium or an aryl-Grignard reagent to a cyclohexanone (e.g. [3]) or a 2-methoxy-2-cyclohexen-1-one [4], followed by more or less tedious transformations of the saturated ring. A retrosynthetic analysis of **a**, however, shows that the precursor **b** might be obtained in a more direct way by Michael addition of a 1,3-cyclohexanedione to a quinone or a quinone acetal (Scheme 1). Encouraging reports [5] have demonstrated that, by the use of quinone monoacetals **c**, the regioselectivity of the 1,4-addition can be controlled, and that the major drawback of the addition to quinones, oxidation of the product by

¹) Part I: [1].

²) These results are part of the planned Ph. D. thesis of U.W.



the starting quinone followed by a second addition, is eliminated. In our special case the acetal protection could also prevent the formation of unwanted tetrahydro-1-dibenzofuranones **d** (*Scheme 1*). Cyclohexanediones suited for the final synthesis of **1** can then be derived from the acid **2**, which is readily available from 3-oxoglutarate [6].

2. Addition of Dione 3 to p-Benzoquinone Mono- and Bisacetals. – Before turning to more complex substrates, the reaction of the parent cyclohexanedione 3 with the acetals of p-benzoquinone was studied. Reaction of dimethyl acetal 4 with dione 3 in



Scheme 2

boiling *t*-BuOH catalyzed by *t*-BuOK yielded 65% of the polycyclic product 5^3) and 18% of the tetrahydrodibenzofuranone 6^4). The analogous process with the ethylene acetal 7 afforded 85% of the adduct 8 as the only isolated product (*Scheme 2*).

All attempts to isolate 2-aryl-3-methoxy-2-cyclohexenones derived from structures of type **b** (*Scheme 1*) by treatment of the crude mixture with $CH_3SO_3H/CH_3OH/ClCH_2CH_2Cl$ (reflux)⁵) resulted in decomposition of all products except the dibenzofuranone **6**.

More successful was the ZnCl₂-catalyzed reaction of **3** with bisacetal **11** according to [8]. *O*-Alkylation ((CH₃O)₂SO₂/K₂CO₃) of the crude mixture containing the dione **12** afforded the target compound **9** in 57% yield together with **6** (21%, *Scheme 2*).

No attempts to extend this transformation to the methoxy-substituted bisacetal 13 were made, since the 4'-methoxy-substituted analogue of 9 (Scheme 2) could be obtained by a different route [9]. Furthermore, bisacetal 13 most probably leads to the regio-isomer resulting from attack at C(6), since the carboxonium ion 14 is the thermodynamically favored intermediate formed from 13 (Scheme 3). The desired adduct (b, $R^3 = OCH_3$, Scheme 1), however, requires bond formation at C(5).

Reaction of the 2-methoxy-substituted monoacetal 15 with dione 3 unfortunately gave an intractable mixture of complex products. Acetal 15 was obtained in low yield (< 10%) by T1(NO₃)₃ oxidation of β -hydroxyethyl phenyl ether 16 in CH₃OH/KHCO₃



³) Analogous products, formed by a double *Michael* addition to the quinone monoacetal, have been obtained with β -ketoesters [5a], β -ketoester-derivatives [5b], and with a *Schiff*'s-base derivative of glycine [5c].

⁴) Inferior results were obtained under the following conditions: EtONa/EtOH (60°), NaH/dioxane (reflux), *t*-BuOK/THF (reflux), or *t*-BuOK/THF/B(OCH₃)₃ (reflux).

⁵) Dione 3, a very polar compound, which is difficult to purify chromatographically on silica gel, can be efficiently O-methylated by this method, which has been developed for the esterification of carboxylic acids [7]. However, it was discovered later, that the 2-aryl-substituted derivative 9 is slowly cleaved to the 6-aryl-5-oxohexanoate 10 under these conditions (Scheme 2).

according to [10]⁶). Phenol **16** was prepared from phosphate **17** (obtained by reduction of 2-methoxy-*p*-benzoquinone (**18**) with P(OCH₃)₃/TMSCl [13]) in *ca*. 75% overall yield by *O*-alkylation with methyl bromoacetate (\rightarrow **19**), phosphate cleavage (CH₃OH/CsF) affording **20**, and LiAlH₄-reduction (*Scheme 3*).

Next, it was examined whether the polycyclic adduct **8**, obtainable in 85% yield, could be transformed to a 2-arylcyclohexanone related to **9** (*Scheme 2*). Attempts to hydrolyze the spiroacetal of **8** by treatment with 6N HCl in dioxane resulted in formation of the dibenzofuranone **21** which was isolated in 91% yield as its acetate **22** (*Scheme 4*). Reaction of **8** with (CH₃O)₂SO₂/K₂CO₃ in boiling acetone afforded 94% of the polyfunctional compound **23**, which corresponds to the primary adduct of dione **3** to monoacetal **7** (*Scheme 4*). Compound **23** was found to be very labile, returning to **8** under very mild conditions⁷), sometimes simply on standing in the refrigerator⁸)⁹.



- ⁶) The same method afforded the corresponding dimethyl acetal from 2,4-dimethoxyphenol in 77% yield. The formation of 2,4-dimethoxyphenol (30%) from the oxidation of 16 implies, that transacetalization with the solvent CH₃OH is a major problem [11]. Since the reaction of 15 with 3 was not successful, no attempts to improve the conversion of 16 to 15 were made. A promising method would be the photochemically mediated oxidation with HgO/l₂ in benzene [12].
- ⁷) Prolonged exposure to silica gel, silica gel/15% H₂SO₄ in CH₂Cl₂ [14], 2% aq. AcOH in THF, *m*-chloroperbenzoic acid in benzene, dipotassium azodicarboxylate in AcOH, Pd/BaCO₃ in AcOEt. Treatment of 23 with TsOH H₂O in boiling acetone or with CH₃SO₃H in Et₂O at r.t., on the other hand, gave 21 in low yield (no exper. description).
- ⁸) The dimethyl acetal corresponding to 23 was obtained analogously from adduct 5 (no exper. description). This compound is, however, even more unstable than 23.
- ⁹) Models of 23 show that this structure is sterically congested, and that the conformer 23a with a pseudoaxial methoxycyclohexenone ring could be favored. This would explain the facile intramolecular *Michael* addition (--8) by the proximity of the C(2')-keto-O-atom to the C(6)=C(7) bond and the reluctance to aromatize by the equatorial position of H--C(10). The ¹H-NMR spectrum (300 MHz) of 23 measured at the probe temp. exhibits line-broadening due to dynamic effects (either rotation around the C(10)-C(1') bond or ring inversion of the spirocyclic moiety). With the exception of the CH₃O-group, this broadening is restricted to the signals of the spirocyclic part. Sharper signals are observed for one of the spiroacetal-Hatoms and for H-C(7). The ¹³C-NMR spectrum (75 MHz) of 23 exhibiting broad lines at the probe temperature is resolved into the spectra of two conformers (ratio ca. 7:3) upon cooling to -20° (see *Exper. Part*).

To explore the synthetic potential of 23, attempts were made to reduce selectively the C(6)=C(7) bond. The desired spiroacetal 24 could be obtained in high yield by hydrogenation with Pd/CaCO₃ in CH₃CN [15]¹⁰). This derivative rearranges under very mild conditions (*e.g.* chromatography on silica gel) to a crystalline derivative with proposed structure 25, which is transformable to 26 by treatment with CH₃SO₃H/ CH₃OH at r.t. (*Scheme 4*)¹¹). Attempted epoxidation of 23 with *m*-chloroperbenzoic acid gave 8 and treatment with H₂O₂/NaOH according to [16] led to intractable mixtures¹²).

3. Reaction of Benzoquinone Monoacetals with Silylenol Ethers. – The inaccessability of 2-arylcyclohexane-1,3-diones **b** by 1,4-addition to quinone monoacetals **c** (Scheme 1) might be due to the harsh conditions which are required to effect the aromatization of the primary adduct **e**, proceeding via an unfavorable carboxonium ion **f** (Scheme 5). The adduct **g**, on the other hand, resulting from the reaction with bisacetal **11** (Scheme 2), is aromatized via the less biased intermediate **h** (Scheme 5). Hoping that the primary Michael adducts to quinone monoacetals could be trapped as the enolethers **i**, we investigated the reactions with silylenol ethers.



Reaction of monoacetal 4 with β -(trimethylsilyloxy)cyclohexenone 27 catalyzed by 0.1 equiv. of Bu₄NF/molecular sieves (m.s.) [17]¹³) at low temperature (-78° - r.t.) gave a low yield (12%) of **6** (no exper. description). The same reactants with 1 equiv. of fluoride afforded, after treatment with CH₃OH/CH₃SO₃H/ClCH₂CH₂Cl⁵) and acetylation, 4% of diacetate 28 in addition to 8% of **6**. At elevated temperature (reflux in THF) the amount of **6** could be increased to 50% (*Scheme 6*). To clarify the reaction path leading to the methylated derivative **6**, silylenol ether **27** was treated under the same conditions with the monosubstituted quinone monoacetals **29**, **30**, **31**, and **32**. *The 2-substituted acetals 29 and 30 are thereby transformed to the 6-substituted 8-metho-xydibenzofuranones 33 (30%) and 34 (16%), clearly demonstrating that 27 is added to the ene-acetal and not to the enone function of the quinone monoacetals (Scheme 6). The hydroquinone monomethyl ethers corresponding to the monoacetals 29 and 30 are*

¹⁰) Other methods for double-bond reduction convert 23 to 8 before hydrogenation⁷). With Pd/BaSO₄ (55 psi of H₂) or ((C₆H₅)₃P)RhCl hemiacetal 25 was isolated, and 10% Pd/C reduced the methoxycyclohexenone moiety to a cyclohexanone.

¹¹) The structures of **25** and **26** are in accordance with the spectroscopic data (see *Exper. Part*). Due to the complexity of the ¹H-NMR spectra, an assignment of the rel. configuration of C(9) was not possible.

¹²) Treatment of the diethyl malonate adduct to quinone monoacetal **4** [5a] according to **16** gave 34% of epoxide at 70% conversion (no exper. description).

¹³) Reaction of 4 and 27 catalyzed by TiCl₄ [18] or CF₃SO₃TMS [19] led to intractable product mixtures.



thereby isolated in considerable amounts $(27\% \text{ and } 41\%, \text{ resp.})^{14}$. Interestingly the 3-substituted monoacetals **31** and **32** are inert to treatment with silylenol ether **27** and fluoride in boiling THF, being recovered in high yield (75–88%, no exper. description)¹⁵).

Finally, the reaction of quinone monoacetal 4 with the more reactive silylenol ether 35 was investigated under a variety of conditions (*Scheme 6*). The desired 2-arylcyclohexanone 36 could be isolated in very low yield (4–9%) together with *ca.* 40% of hydroquinone monomethyl ether isolated as acetate 37 from reaction mixtures obtained with an excess of CF₃SO₃TMS/Et₃N [19]. Fluoride catalysis (0.1 equiv. of Bu₄NF, $-78^{\circ} - r.t.$) gave complex mixtures containing 6–18% of the tricyclic monoacetal 38, resulting from a double 1,4-addition to 4¹⁶). When quinone monoacetal 4 and enol ether 35 were heated to 200–210° without solvent¹⁷), hydroquinone monomethyl ether was isolated in quantitative yield as its acetate 37. A similar result (61% of 37) was obtained with TiCl₄-catalysis [18]. This implies that reactions of quinone acetals initiated by single-electron transfer, so far observed with cuprate and alkyl-lithium reagents [21], can also occur with silylenol ethers¹⁸).

¹⁴) Compound 6 isolated in the base-catalyzed reaction of monoacetal 4 and dione 3 is most probably formed by a similar mechanism.

¹⁵) Judged by the results of *Coates & MacManus* [5b] and by *Parker et al.* [5c], the monoacetals **31** and **32** are expected to react with dione **3** under base catalysis, giving polycyclic products of type **5** (*Scheme 2*), which should be further transformable according to *Scheme 4*. Base-catalyzed reaction of **3** with the monoacetals **29** and **30**, on the other hand, is most probably not successful (in analogy to **15**, *Scheme 3*).

¹⁶) Another complex product (4-10%) originating from 2 equiv. of acetal 4 and 1 equiv. of cyclohexanone (mol. wt. 356) could be isolated. Its complex structure, however, could not be elucidated by routine spectroscopy.

¹⁷) At elevated temp. Michael addition of (2-tetrahydropyranyl)enol ethers to enones has been observed [20].

¹⁸) While the TiCl₄-catalyzed reaction of **4** and **35** proceeds at low temperature, the uncatalyzed process as monitored by ¹H-NMR starts not below 180°.

4. Conclusions. – The direct access to 2-arylcyclohexane-1,3-diones and 2-arylcyclohexanones by 1,4-addition to quinone monoacetals failed; the desired products were formed at best in very low yields. It is still possible that these difficulties, which are in part due to steric effects, could be overcome by working at high pressure [22] or by using tris(dimethylamino)sulfonium difluorotrimethylsiliconate as catalyst [23]. More promising is the reaction of 3 with bisacetal 11, giving the desired product 9 in reasonable yield. Preparatively rewarding is the high yield of the polycyclic compound 8 and its facile transformation to the methyl-enol ether 23 and to the dibenzofuranone 21, which is thereby obtained in much better yields, than from reactions of p-benzoquinone and dione 3 (cf. [9]). Since it was recently found that dibenzofuranone derivatives d can be efficiently transformed to enol ether derivatives of 2-arylcyclohexane-1,3-diones b [9], the synthetic goal depicted in *Scheme 1* could still be attained in an indirect way.

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Experimental Part

General. See [24]. Purification of Solvents. THF, distillation from Na/benzophenone; CH_2Cl_2 , filtration through Alumina (Woelm, basic, activity I); CH_3OH , distillation from Mg(OCH₃)₂ formed by the addition of Mg-turnings. Materials: The quinone bisacetal **11** was prepared electrolytically [25]. The quinone monoacetals **4**, **7**, and **32** were obtained by regioselective monohydrolysis of the corresponding bisacetals [26], which have been prepared electrolytically [25]. Oxidation of the corresponding hydroquinone monomethyl ethers with Tl(NO₃)₃/KHCO₃ in CH₃OH according to [10] afforded the monoacetals **30** and **31**. The silylenol ethers **27** and **35** were obtained according to [27] and [28], respectively. Bayer-Villiger oxidation of 2,4-dimethoxybenzaldehyde according to [29] at r.t. (boiling under reflux as described in [29] is not needed) gave, after hydrolysis of the formate, 2,4-dimethoxybenol.

1. 2,4,4-Trimethoxy-2,5-cyclohexadien-1-one (29). – To a solution of 2,4-dimethoxyphenol (see above; 87 mg, 0.564 mmol) in dry CH₃OH (6 ml), KHCO₃ (617 mg, 6.16 mmol) and, after cooling to -5° (ice/NaCl), a solution of T1(NO₃)₃ (240 mg, 0.62 mmol) in dry CH₃OH (5 ml) were added. After stirring for 22 min, the mixture was added to ice-cooled sat. NaHCO₃-soln., filtered (*Celite*), and worked up with Et₂O (addition of ice). Chromatography (alumina, basic, activity III, CH₂Cl₂) gave 80 mg (77%) of **29**, m. p. 135–136° (Et₂O/hexane). IR (CHCl₃): 3035w, 3005m, 2970m, 2940m, 2910w, 2835m, 1686s, 1655s, 1650s, 1625m, 1597m, 1460m, 1443w, 1372m, 1310w, 1255m, 1175m, 1120m, 1085s, 1040m, 1020m, 992w, 951s, 910w, 878m, 843m. ¹H-NMR (80 MHz, CDCl₃): 3.37 (s, 2 CH₃O-C(4)); 3.70 (s, CH₃O-C(2)); 5.69 (d, J = 3, H-C(3)); 6.23 (d, J = 10, H-C(6)); 6.83 (dd, J = 10 and 3, H-C(5)). MS: 184 (26, M^+), 169 (36), 154 (86), 153 (100), 141 (14), 139 (66), 138 (21), 126 (7), 125 (42), 123 (10), 111 (35), 110 (23), 108 (30), 95 (19), 82 (15), 79 (14), 69 (52), 55 (12), 54 (18), 53 (24), 52 (27), 51 (16), 41 (10), 39 (18). Anal. calc. for C₉H₁₂O₄ (184.18): C 58.69, H 6.58, O 34.74; found: C 58.69, H 6.57, O 34.90.

2. Preparation of the Cyclic Acetal 15. -2.1. Methyl [4-(Dimethylphosphoryloxy)-3-methoxyphenoxy]-acetate (19). A mixture of phosphate 17 [13] (116 mg, 0.458 mmol), KHCO₃ (640 mg, 4.64 mmol), and methyl bromoacetate (60 µl, ca. 0.6 mmol) in acetone (7 ml) was boiled for 46 h under reflux (Ar). Filtration (*Celite*) and chromatography (15 g of silica gel, CH₂Cl₂/CH₃OH 96:4) of the residue obtained by evaporation of the filtrate afforded 144 mg (96%) of 19. IR (CHCl₃): 3030w, 3000m, 2955m, 2855w, 2830w, 1756s, 1611m, 1600m, 1505s, 1464m, 1450m, 1436m, 1374w, 1276s, 1182s, 1163s, 1129m, 1112w, 1050s, 956s, 938m, 920m, 856s, 839w. ¹H-NMR (80 MHz, CDCl₃): 3.80 and 3.84 (2s, OCH₂COOCH₃, CH₃O-C(3)); 3.86 (d, J = 11, (CH₃O₂PO₂-C(4)); 4.59 (s, OCH₂COOCH₃); 6.34 (dd, J = 9 and 3, H-C(6)); 6.61 (d, J = 3, H-C(2)); 7.16 (dd, J = 9 and ≈ 1 , H-C(5)). MS (di.): 320 (70, M^+), 289 (3), 261 (7), 247 (13), 231 (12), 194 (9), 135 (18), 127 (66), 109 (100), 93 (9), 79 (18), 45 (14).

2.2. Methyl (4-Hydroxy-3-methoxyphenoxy)acetate (20). A solution of 19 (140 mg, 0.438 mmol) and CsF (916 mg, 6 mmol) in CH₃OH (4 ml) was stirred for 68 h at r. t. (Ar). Workup with Et₂O gave 89 mg (96%) of 20. IR (CCl₄): 3560m, 3600–3200w, 3005w, 2950m, 2935w, 2845w, 1770s, 1742s (in CHCl₃ one band at 1755s), 1623m, 1612w, 1505s, 1465m, 1451m, 1435m, 1380m, 1352w, 1264m, 1229s, 1200s, 1156s, 1122m, 1106w, 1075m, 1034m, 984w, 940w, 925m, 835m. ¹H-NMR (80 MHz, CDCl₃): 3.78 and 3.83 (2s, OCH₂COOCH₃, CH₃O-C(3)); 4.56 (s, OCH₂COOCH₃); 5.40 (br., $w_{1/2} \approx 3$, HO-C(4)); 6.33 (dd, J = 9 and ≈ 3 , H-C(6)); 6.58 (d, $J \approx 3$, H-C(2)); 6.79 (d, J = 9, H-C(5)). MS: 212 (83, M^+), 197 (3), 153 (13), 139 (100), 123 (15), 111 (20), 110 (21), 109 (10), 95 (6), 79 (13), 74 (12), 45 (11).

2.3. 4-(2'-Hydroxyethoxy)-2-methoxyphenol (16). A solution of 20 (1.708 g, 8.06 mmol) in dry Et₂O (20 ml) was added dropwise to 1.533 g (40.3 mmol) of LiAlH₄ in Et₂O (30 ml). After stirring for 14 h at r.t. (Ar), *Celite* was added and the reaction was quenched by the addition of sat. (NH₄)₂SO₄-soln. (dropwise, until the precipitated solids coagulate with the *Celite*). Filtration (*Celite*) and chromatography (100 g of silica gel, cyclohexane/CH₂Cl₂/AcOEt 1:1:2) of the residue of the filtrate yielded 1.20 g (81%) of 16, m.p. 65-66° (Et₂O). IR (CHCl₃): 3595m, 3550s, 3600-3200m, 3030w, 3000w, 2960m, 2935m, 2870w, 2845w, 1622m, 1610m, 1502s, 1463m, 1449s, 1431m, 1377s, 1355m, 1300w, 1287w, 1264m, 1190s, 1156s, 1119m, 1106w, 1070m, 1029s, 955m, 895m, 865w, 835m, 820w. ¹H-NMR (80 MHz, CDCl₃): 2.1-2.6 (br., HOCH₂CH₂O-C(4)); 3.82 (s, CH₃O-C(2)); 3.6-4.1 (m, HOCH₂CH₂O-C(4)); 5.2-5.6 (br., HO-C(1)); 6.37 (dd, J = 8 and 3, H-C(5)); 6.50 (d, $J \approx 3$, H-C(3)); 6.80 (d, J = 8, H-C(6)). MS: 184 (52, M^+), 153 (2), 140 (100), 125 (73), 111 (8), 97 (26), 85 (15), 83 (22), 52 (10), 45 (12).

2.4. 7-Methoxy-1,4-dioxaspiro[4.5]deca-6,9-dien-8-one (15). A solution of Tl(NO₃)₃ (600 mg, 1.5 mmol) in dry CH₃OH (5 ml) was added slowly to a ice-cooled solution of 16 (240 mg, 1.302 mmol) in dry CH₃OH (15 ml) containing 1.409 g (14 mmol) of powdered KHCO₃ (Ar). After stirring for 1 h at 0°, the reaction was quenched by the addition of ice-cooled sat. NaHCO₃-soln. Filtration (*Celite*), workup with Et₂O and chromatography (20 g of silica gel, cyclohexane/CH₂Cl₂/AcOEt 12:8:5) gave 61 mg (30% of 2,4-dimethoxyphenol, 20 mg (8%, 10% based on consumed 16) of 15, and 52 mg (21%) of starting material 16. IR (CHCl₃): 3030w, 3010w, 2960w, 2935w, 2890m, 2845w, 1685s, 1650s, 1623s, 1665w, 1460m, 1455m, 1442w, 1368s, 1246m, 1180m, 1118s, 1085s, 1020s, 985w, 960s, 945s, 880w, 836m. ¹H-NMR (80 MHz, CDCl₃): 3.71 (*s*, CH₃O-C(7)); 4.18 (*s*, 2H-C(2), 2H-C(3)); 5.50 (*d*, $J \approx 3$, H-C(6)); 6.15 (*d*, J = 10, H-C(9)); 6.68 (*dd*, J = 10 and ≈ 3 , H-C(10)). MS: 182 (82, M^{-1}), 167 (8), 154 (53), 152 (12), 151 (10), 139 (59), 126 (20), 122 (52), 113 (24), 109 (34), 98 (45), 95 (53), 94 (16), 79 (100), 69 (59), 66 (14), 64 (14), 53 (29), 51 (55), 50 (22), 43 (12), 39 (26).

3. Reaction of 1,3-Cyclohexanedione (3) with p-Benzoquinone Acetals. - 3.1. Reaction with Dimethyl Monoacetal 4. A solution of 3 (495 mg, 4.42 mmol), monoacetal 4 (500 mg, 3.24 mmol), and t-BuOK (158 mg, 1.4 mmol) in t-BuOH (25 ml) was boiled under reflux for 4 d (Ar). Chromatography (100 g of silica gel, cyclohexane/CH₂Cl₂/AcOEt 1:2:2) of the residue, obtained by evaporation, gave 129 mg (18%) of dibenzo-furane 6 and 567 mg (65%) of 13,13-dimethoxy-8-oxatricyclo[7.3.1.0^{2.7}]tridec-2(7)-ene-3,11-dione (5). IR (CHCl₃): 3030w, 2995m, 2960m, 2945m, 2910w, 2870w, 2835m, 1720s, 1652s, 1620s, 1455m, 1428w, 1412w, 1387s, 1369s, 1357m, 1326w, 1293m, 1280m, 1265m, 1185s, 1148m, 1135m, 1121s, 1100w, 1071s, 1064s, 1038s, 1021s, 1004m, 995w, 969m, 930w, 897m, 871w. ¹H-NMR (300 MHz, CDCl₃): 1.85-2.05 (m, 5 main peaks, 2H-C(5)); 2.3-2.4 (m, 2H-C(4), 2H-C(6)); 2.44 (dt, J = 16 and 2.5) and 2.67 (dd, J = 16 and 4.5) (2H-C(12)); 2.59 (ddd, J = 17, 2.5, and 2) and 2.80 (dd, J = 17 and 4.5) (2H-C(10)); 3.29 and 3.39 (2s, 2 CH₃O-C(13)); 3.35-3.45 (m, H-C(1)); 4.65 (ddd, J = 4.5, 3.5, and 2, H-C(9)). MS: 266 (50, M⁺), 238 (18), 235 (11), 234 (10), 219 (8), 210 (38), 195 (22), 191 (16), 163 (12), 151 (24), 101 (100), 91 (12), 77 (10), 65 (9), 55 (49), 43 (41), 41 (13), 39 (11).

Methoxy-1,2,3,4-tetrahydrodibenzofuran-1-one (6). IR (CHCl₃): 3030w, 2990w, 2950m, 2885w, 2830m, 1660s, 1615m, 1584s, 1477s, 1455s, 1429s, 1395s, 1355w, 1325w, 1312w, 1271s, 1169s, 1151s, 1133m, 1102w, 1055m, 1040w, 1025m, 1005s, 893w, 880w, 870w, 850m, 824w, 635w. ¹H-NMR (100 MHz, CDCl₃): 2.1–2.4 (m, 2H–C(3)); 2.45–2.65 and 2.85–3.1 (2m, each 3 main peaks, 2H–C(2), 2H–C(4)); 3.81 (s, CH₃O–C(8)); 6.84 (dd, J = 9 and 3, H–C(7)); 7.30 (d, J = 9, H–C(6)); 7.49 (d, J = 3, H–C(9)). MS: 216 (100, M^+), 201 (5), 189 (12), 188 (99), 173 (7), 160 (72), 145 (6), 132 (31), 117 (9), 115 (10), 89 (20), 63 (17), 43 (10). Anal. calc. for C₁₃H₁₂O₃ (216.23): C 72.21, H 5.59; found: C 72.23, H 5.58.

3.2. {8-Oxatricyclo[7.3.1.0^{2.7}]tridec-2(7)-ene-3,11-dione}-13-spiro-2'-{1',3'-dioxolane} (8). A mixture of dione 3 (1.338 g, 11.95 mmol), cyclic monoacetal 7 (1.311 g, 8.63 mmol), and t-BuOK (337 mg, 3 mmol) in t-BuOH (50 ml) was boiled under reflux for 23 h (Ar). Evaporation (aspirator) and chromatography (200 g of silica gel, cyclohexane/CH₂Cl₂/AcOEt 1:2:2) of the residue gave 1.954 g (85% based on 7) of 8, m.p. 156–157° (Et₂O/hexane). UV (EtOH): 261 (11000). IR (CHCl₃): 3030w, 3000m, 2955m, 2895m, 2840w, 1720s, 1654s,

1620s, 1473w, 1455w, 1429w, 1415w, 1390s, 1375s, 1356w, 1340w, 1325w, 1294w, 1270m, 1187s, 1150m, 1131s, 1092m, 1063s, 1025s, 1005m, 987w, 971w, 949m, 930w, 906w, 873w, 858w. ¹H-NMR (300 MHz, CDCl₃): 1.9–2.02 (*m*, 5 main peaks, 2H–C(5)); 2.3–2.48 (*m*, 2H–C(4), 2H–C(6)); 2.48 (*ddd*, J = 16, 3, and 2.5) and 2.93 (*dd*, J = 16 and 4.5) (2H–C(12)); 2.65 (*dt*, J = 17 and 2.5) and 2.91 (*dd*, J = 17 and 4.5) (2H–C(10)); 3.07–3.14 (*m*, 4 main peaks, H–C(1)); 4.06–4.2 (*m*, 2H–C(4'), 2H–C(5')); 4.34 (*ddd*, J = 4.5, 3, and 2.5, H–C(9)). ¹³C-NMR (25 MHz, CDCl₃): 20.7 (C(5)); 27.6 and 36.1 (C(4), C(6)); 33.2 (C(1)); 43.2 and 45.8 (C(10), C(12)); 65.4 and 65.6 (C(4'), C(5')); 75.9 (C(9)); 103.2 (C(2)); 113.0 (C(13)); 168.8 (C(7)); 196,1 (C(3)); 206.9 (C(11)). MS: 264 (57, M^+), 236 (17), 209 (10), 208 (46), 164 (18), 163 (11), 136 (15), 108 (11), 99 (100), 77 (7), 57 (34), 56 (16), 55 (53), 43 (31), 42 (14), 41 (31), 39 (15). Anal. calc. for C₁₄H₁₆O₅ (264.28): C 63.63, H 6.10, O 30.27; found: C 63.86, H 6.22, O 30.50.

3.3. Reaction of Dione 3 with Bisacetal 11. - 3.3.1. 2-(2',5'-Dimethoxyphenyl)-3-methoxy-2-cyclohexen-1-one (9). A solution of $ZnCl_2$ (1.04 g, 7.63 mmol, dried by melting at h.v.) and 3 (3.1 g, 27.64 mmol) in dry dioxane (20 ml) was heated to the boiling point. Solvent was removed continuously by distillation while a solution of bisacetal 11 (2.64 g, 13.18 mmol) in dioxane (5 ml) was added (25 min). Within 2 h, most of the volatile components of the mixture were evaporated at normal pressure, and the residue was worked up with CH₂Cl₂. The crude mixture was treated with $(CH_3O)_2SO_2$ (3.8 ml, ca. 39.9 mmol) and powdered K_2CO_3 (23.3 g) in acetone (500 ml) for 24 h (stirring at r.t.). Filtration (Celite), evaporation, and chromatography (200 g of silica gel, cyclohexane/CH₂Cl₂/AcOEt 1:2:2) afforded 602 mg (21% based on 11) of 6 (see 3.1), 525 mg (15% based on 3) of 3-methoxy-2-cyclohexen-1-one, and 1.98 g (57% based on 11) of 9, m.p. 92-93° (AcOEt/pentane). IR (CHCl₃): 3030w, 2995m, 2945s, 2905w, 2870w, 2850w, 2830m, 1645s, 1605s, 1491s, 1457s, 1408m, 1365s, 1355s, 1329w, 1309m, 1289w, 1271s, 1180w, 1170m, 1158m, 1085w, 1080s, 1037s, 1010w, 1000m, 960w, 935w, 916w, 908w, 883m, 855w. ¹H-NMR (80 MHz, CDCl₃): 1.85–2.3 (m, 2H–C(5)); 2.3–2.9 (m, 2H–C(4), 2H–C(6)); 3.65, 3.71, and 3.75 (3s, CH₃O-C(3), CH₃O-C(2'), CH₃O-C(5')); 6.55-6.7 (m, 1H) and 6.75-6.95 (m, 2H) (H-C(3'), H-C(4'), H-C(6')). ¹³C-NMR (75 MHz, CDCl₃): 20.6 (C(5)); 25.9 and 36.7 (C(4), C(6)); 55.5, 55.9, and 56.4 (3 CH₃O); 112.2, 113.1, and 117.9 (C(3'), C(4'), C(6')); 116.8 and 124.2 (C(2), C(1')); 151.6 and 153.1 (C(2'), C(5')); 173.0 (C(3)); 196.9 (C(1)). MS: 262 (77, M⁺), 247 (2), 231 (100), 219 (15), 191 (9), 187 (15), 161 (12), 131 (9), 115 (6), 91 (8), 84 (9), 55 (10). Anal. calc. for $C_{15}H_{18}O_4$ (262.30): C 68.68, H 6.92, O 24.40; found: C 68.55, H 6.99, O 24.78.

3.3.2. Methyl 6-(2',5'-Dimethoxyphenyl)-5-oxohexanoate (10). A solution of 9 (111 mg, 0.424 mmol) in 5 ml of a mixture of CH₃OH (10 ml), ClCH₂CH₂Cl (20 ml), and CH₃SO₃H (0.1 ml) [7] was boiled under reflux for 23 h. Workup with AcOEt and chromatography (10 g of silica gel, cyclohexane/CH₂Cl₂/AcOEt 1:2:2) gave 30 mg (25%) of 10 and 21 mg (19%) of 9. IR (CCl₄): 3000m, 2950m, 2905m, 2830m, 1740s, 1713s, 1592w, 1500s, 1463m, 1433m, 1370-1360w, 1311m, 1280m, 1265w, 1224s, 1201w, 1193w, 1180m, 1158m, 1128w, 1090w, 1050s, 1029m, 935w, 880w, 855w, 710m, 702m. ¹H-NMR (300 MHz, CDCl₃): 1.8-2.0 (m, 5 main peaks, 2H–C(3)); 2.25–2.35 and 2.45–2.55 (2m, each 3 main peaks, 2H–C(2), 2H–C(4)); 3.62 (s, 2H–C(6)); 3.64, 3.751, and 3.754 (3s, 3 OCH₃); 6.71 (dd, $J \approx 2.5$ and 1, H–C(6')); 6.76 (dd, J = 9 and 2.5, H–C(4')); 6.79 (dd, J = 9 and 1, H–C(3')). ¹³C-NMR (75 MHz, CDCl₃): 19.0 (C(3)); 33.0 and 40.7 (C(2), C(4)); 44.9 (C(6)); 51.5, 55.7, and 55.9 (3 CH₃O); 111.4, 112.7, and 117.4 (C(3'), C(4'), C(4'), C(4), 124.6 (C(1')); 151.6 and 153.5 (C(2'), C(5')); 173.6 (C(1)); (97), 97 (10), 91 (24), 77 (17), 65 (15), 59 (55), 55 (26), 51 (7), 41 (10).

4. Transformations of Adduct 8. – 4.1. *1-Oxo-1,2,3,4-tetrahydrodibenzofur-8-yl Acetate* (22). A solution of **8** (35.4 mg, 0.134 mmol) in dioxane (4 ml) and 6N aq. HCl (2 ml) was stirred at r.t. for 5 d. Evaporation at reduced pressure, acetylation (Ac₂O/pyridine, 0.5 ml of each, over night), evaporation, and chromatography (10 g of silica gel, cyclohexane/CH₂Cl₂/AcOEt 1:2:2) gave 30 mg (91%) of **22**, m. p. 110–111° (Et₂O/hexane). IR (CHCl₃): 3030w, 2998w, 2955m, 2890w, 2870w, 2835w, 1755s, 1665s, 1622w, 1588s, 1451s, 1429w, 1395s, 1367s, 1325w, 1185m, 1166s, 1141m, 1130m, 1106w, 1055m, 1042w, 1006s, 934w, 920m, 896m, 883m, 865w. ¹H-NMR (80 MHz, CDCl₃): 2.05–2.45 (m, 2H–C(3)); 2.30 (s, CH₃COO–C(8)); 2.45–2.75 and 2.9–3.2 (2m, 2H–C(2), 2H–C(4)); 7.00 (dd, J = 9 and \approx 3, H–C(7)); 7.43 (d, J = 9, H–C(6)); 7.74 (d, $J \approx$ 3, H–C(9)). MS: 244 (8, M^+), 202 (100), 175 (8), 174 (74), 146 (26), 118 (11), 89 (10), 63 (9), 55 (6), 43 (22).

4.2. 10-(2'-Methoxy-6'-oxo-1'-cyclohexenyl)-1,4-dioxaspiro[4.5]dec-6-en-8-one (23). A mixture of 8 (122.7 mg, 0.464 mmol), (CH₃O)₂SO₂ (66 µl, ca. 0.6 mmol), and powdered K₂CO₃ (383 mg) in acetone (10 ml) was boiled under reflux for 15 h. Filtration (*Celite*), evaporation, and chromatography (15 g of silica gel, cyclohexane/CH₂Cl₂/AcOEt 1:2:2) yielded 122 mg (94%) of 23. IR (CHCl₃): 3035w, 2985m, 2945m, 2885m, 2850w, 1675s, 1640s, 1608s, 1456m, 1405w, 1374s, 1345m, 1325w, 1277m, 1256s, 1163s, 1140m, 1120m, 1088m, 1059w, 1038m, 1024m, 1002w, 948m, 932w, 915m. ¹H-NMR (300 MHz, CDCl₃): 1.9–2.1 (m, 2H–C(4')); 2.25–2.5 and

2.5–2.7 (2m, 2H–C(3'), 2H–C(5')); 2.6–2.95 (m, 2H–C(9)); 3.63 (br., $w_{Y_2} \approx 15$, CH₃O–C(2')); ≈ 3.5 –4.2 (m, 2H–C(2), 2H–C(3), H–C(10)); 6.03 (d, J = 10, H–C(7)); 6.46 (d, J = 10, further broadened, $w_{Y_2} \approx 5$, H–C(6)). ¹³C-NMR (75 MHz, CDCl₃, –20°, multiplicities determined by gated spin-echo techniques, mixture of two conformers, signals of minor conformer (20–30%) in parentheses): 20.3 (C(4')); 25.6 (25.4), 36.2 (37.6?), and 39.2 (39.6) (C(9), C(3'), C(5')); 36.7 (37.5) (C(10)); 54.4 (55.8) (OCH₃); 64.9 (65.0) and 66.2 (65.9) (C(2), C(3)); 104.8 (105.4) and 115.6 (113.9) (C(5), C(1')); 130.5 (130.0) (C(7)); 146.3 (146.0) (C(6)); 173.9 (175.8) (C(2')); 198.0 (198.2) and 199.0 (199.6) (C(8), C(6')). MS (di.): 278 (6, M^+), 164 (2), 250 (1), 236 (5), 208 (2), 151 (2), 126 (100), 103 (5), 99 (11), 98 (53), 55 (11).

4.3. 6-(2'-Methoxy-6'-oxo-1'-cyclohexenyl)-1.4-dioxaspiro[4.5]decan-8-one (24). Freshly prepared enone 23 (49.8 mg, 0.179 mmol) was hydrogenated at normal pressure with 10% Pd/CaCO₃ (16 mg) in CH₃CN (5 ml) for 90 min. Filtration (*Celite*) and evaporation at reduced pressure gave 46 mg (91%) of 24. IR (CCl₄): 2940s, 2890m, 2847w, 2815w, 1708s, 1655s, 1617s, 1455m, 1442m, 1427w, 1415w, 1403m, 1374s, 1343m, 1332m, 1312w, 1287m, 1268s, 1253s, 1235s, 1210w, 1191m, 1168s, 1140s, 1110s, 1079m, 1029s, 980s, 945m, 930s, 920s, 860w. ¹H-NMR (300 MHz, CDCl₃): 1.65–1.85 (m, 1H); 1.9–2.1 (m, 2H–C(4')); 2.17 (td, J = 13 and 6, 1H); 2.25–2.85 (m, 8H); 3.62 (br., $w_{1/2} \approx 5$, CH₃O); 3.7–4.25 (m, 2H–C(2), 2H–C(3), H–C(6)). MS (di.): 280 (31, M^+), 265 (4), 194 (4), 181 (11), 166 (11), 139 (9), 138 (8), 100 (29), 99 (100), 87 (10), 55 (32), 41 (6).

4.4. 11-Hydroxy-2,12,15-trioxatetracyclo[9.4.2.0^{1.9}, 0^{3.8}]heptadec-3(8)-en-7-one (**25**). Crude **23**, freshly prepared from **8** (103 mg, 0.39 mmol) as described in 4.2, was dissolved in benzene/EtOH (10 ml of each) and hydrogenated at normal pressure with ((C_6H_5)₃P)₃RhCl (96 mg) for 20 h. Evaporation and chromatography (10 g of silica gel, CH₂Cl₂/CH₃OH 96:4) gave 82 mg (78% based on **8**) of **25**, n.p. 177–178° (CHCl₃/hexane). UV (EtOH): 263 (14600). IR (CHCl₃): 3580m, 3500–3200w, 2997m, 2950m, 2890m, 1640s, 1610s, 1441m, 1429m, 1387s, 1369m, 1352m, 1332m, 1315w, 1300m, 1270w, 1163s, 1131s, 1113s, 1063m, 1032m, 1015s, 1001w, 968s, 946m, 935s, 903m, 866w, 855w, 835m. ¹H-NMR (300 MHz, CDCl₃): 1.45–1.65 (m, 2H); 1.75–1.25 (m, 5H); 2.2-2.6 (m, 4H); 2.34 (dd, J = 12.5 and 3, H–C(10)); 3.22 (m, $w_{Y_2} \approx 7$, H–C(9)); 3.29 (br., $w_{Y_2} \approx 2$, exchangeable with D_2O , HO–C(11)); 3.85–4.0 (m, 2H), 4.0–4.13 (m, 1H), and 4.13–4.3 (m, 1H) (2H–C(13), 2H–C(14)). ¹³C-NMR (75 MHz, CDCl₃, *: assignment interchangeable): 21.0 (C(5)); 28.6 and 36.4* (C(4), C(6)); 28.6, 32.9 and 36.5* (C(10), C(16), C(17)); 32.6 (C(9)); 64.1 and 64.7 (C(13), C(14)); 101.2 (C(8)); 109.8 and 112.6 (C(1), C(11)); 173.4 (C(3)); 196.9 (C(7)). MS (di.): 266 (66, M^{\pm}), 248 (1), 221 (7), 167 (6), 162 (9), 142 (27), 137 (14), 113 (6), 100 (47), 99 (100), 87 (31), 86 (65), 55 (28), 43 (12), 42 (15).

4.5. 5a, 8, 8-Trimethoxy-1,2,3,4,5a,6,7,8,9,9a-decahydrodibenzofuran-1-one (**26**). A solution of **25** (10.8 mg, 0.041 mmol) and CH₃SO₃H (10 µl) in CH₃OH (2 ml) was stirred at r.t. for 16 h. The reaction was quenched with sat. NaHCO₃-soln. and worked up with Et₂O. Chromatography (10 g of silica gel, CH₂Cl₂/CH₃OH 96:4) gave 10.5 mg (ca. 90%) of **26**. IR (CCl₄): 2940s, 2890m, 2850w, 2835m, 1655s (in CHCl₃: 1645s), 1615s (in CHCl₃: 1608s), 1451m, 1435w, 1429m, 1383s, 1370s, 1353w, 1340w, 1330m, 1291m, 1285m, 1266w, 1247m, 1220w, 1185s, 1170s, 1150s, 1118s, 1063s, 1046s, 1010m, 980m, 950m, 930s, 900m, 865w, 854w, 828s, 707w, 620m. ¹H-NMR (300 MHz, CDCl₃): 1.38–1.55 (m, 1H); 1.68–1.9 (m, 3H); 1.9–2.1 (m, 3H); 2.08 (dd, J = 12.5 and 3, H–C(9)); 2.2–2.37 (m, 1H); 2.37–2.57 (m, 3H); 3.22, 3.23, and 3.36 (3s, 3 CH₃O); 3.46 (m, $w_{Y_{4}} \approx 8$, H–C(9a)). ¹³C-NMR (25 MHz, CDCl₃): 21.0 (C(3)); 25.7, 28.6, 28.9, 34.0, and 36.8 (C(2), C(4), C(6), C(7), C(9)); 31.4 (C(9a)); 47.9, 48.7, and 49.9 (3 CH₃O); 102.2, 103.6, and 113.1 (C(5a), C(8), C(9b)); 172.9 (C(4a)); 195.8 (C(1)). MS (di.): 282 (4, M^+), 251 (23), 250 (43), 235 (14), 219 (26), 181 (100), 162 (8), 153 (11), 138 (8), 101 (16), 89 (21), 88 (94), 58 (24), 55 (20), 43 (20).

5. Reactions of 3-(Trimethylsilyloxy)-2-cyclohexen-1-one (27) with Quinone Monoacetals. – 5.1. Reaction with Monoacetal 4. – 5.1.1. Reaction at r.t. in the Presence of Molecular Sieves (m.s.). A suspension of m.s. 4 Å (2 g) in a solution of (Bu)₄N⁺F⁻·3H₂O (662 mg, 2.1 mmol) in dry THF (5 ml) was kept at r.t. for 20 h. After the addition of 27 (387 mg, 2.1 mmol) and 4 (308 mg, 2 mmol) dissolved in THF (2 ml), the mixture was stirred at r.t. for 5 d. The residue obtained by filtration (*Celile*) and evaporation was heated with 5 ml of a mixture of CH₃OH (10 ml), 1,2-dichloroethane (20 ml) and CH₃SO₃H (0.1 ml) [7] for 20 h under reflux. Evaporation, acetylation (Ac₂O/pyridine over night), evaporation, and chromatography (100 g of silica gel, cyclohexane/CH₂Cl₂/AcOEt 1:2:2) yielded 35 mg (8%) of 6 (see 3.1) and 29 mg (4.5%) of 2-(5'-acetoxy-2'-methoxyphenyl)-3-axo-1-cyclohexenyl acetate (28). IR (CHCl₃): 3040w, 2995w, 2955m, 2873w, 2835w, 1757s, 1675s, 1647m, 1604w, 1583w, 1490m, 1465w, 1455w, 1415w, 1368m, 1354m, 1290w, 1268m, 1178s, 1156s, 1057w, 1034m, 1000s, 980w, 911m, 891w, 872w. ¹H-NMR (100 MHz, CDCl₃): 1.95 and 2.11 (2s, 2 CH₃COO); 2.0–2.35 (m, 2H--C(5)); 2.45-2.9 (m, 2H--C(4)); 7.04 (d, J = 9, H-C(3)). ¹³C-NMR (25 MHz, CDCl₃): 20.8 and 21.0 (C(5), 2 CH₃COO); 2.90 and 37.4 (C(4), C(6)); 55.6 (CH₃O); 114.7 and 115.8 (C(3'), C(4')); 123.1 (C(6')); 125.3 and 126.1 (C2), C(1')); 141.9

(C(5')); 156.7 (C(2')); 166.2, 167.7, and 169.3 $(C(1), 2 CH_3COO);$ 196.8 (C(3)). MS: 318 $(6, M^+),$ 276 (5), 258 (4), 235 (13), 234 (85), 217 (26), 216 (100), 192 (19), 174 (32), 161 (18), 137 (8), 91 (5), 55 (6), 43 (20).

5.1.2. Reaction in Boiling THF. To a solution of $(Bu)_4N^+F^- \cdot 3H_2O$ (26 mg, 0.09 mmol) and 27 (187 mg, 1.014 mmol) in dry THF (2 ml) a solution of 4 (125 mg, 0.81 mmol) in THF (10 ml) was added. After boiling under reflux for 97 h, the solvent was evaporated at reduced pressure, and the residue was boiled under reflux for 43 h in 8 ml of a mixture of CH₃OH (10 ml), ClCH₂CH₂Cl (20 ml) and CH₃SO₃H (0.1 ml) [7]. Chromatography (35 g of silica gel, CH₂Cl₂/CH₃OH 96:4) of the crude material obtained by evaporation gave 88 mg (50% based on 4) of 6 (see 3.1).

5.2. Reaction with the 2-Methoxysubstituted Monoacetal **29**. A solution of **29** (351 mg, 1.905 mmol), **27** (374 mg, 2.03 mmol) and $(Bu)_4N^+F^-\cdot 3H_2O$ (64 mg, 0.2 mmol) in 12 ml of dry THF was boiled under reflux for 5 d. After evaporation of solvent (aspirator) the residue was treated with 7 ml of a mixture of CH₃OH (10 ml), CICH₂CH₂Cl (20 ml) and CH₃SO₃H (0.1 ml) [7] for 24 h under reflux. Evaporation and chromatography (60 g of silica gel, Et₂O/hexane 2:1) gave 80 mg (27%) of 2,4-dimethoxyphenol and 142 mg (30%) of 6,8-dimethoxy-1,2,3,4-tetrahydrodibenzofuran-1-one (33): m.p. 154-155° (Et₂O/hexane). IR (CHCl₃): 3030w, 3000w, 2960m, 2835w, 1665s, 1625s, 1604s, 1590m, 1492s, 1462w, 1453m, 1445w, 1432m, 1405w, 1360w, 1345m, 1318m, 1301s, 1186w, 1176s, 1153s, 1116w, 1059m, 1040m, 1020w, 1005s, 935w, 903w, 885w, 884m, 610m. ¹H-NMR (80 MHz, CDCl₃): 2.1-2.45 (m, 2H-C(3)); 2.45-2.75 and 2.85-3.3 (2m, 2H-C(2), 2H-C(4)); 3.75 and 3.97 (2s, CH₃O-C(6), CH₃O-C(8)); 6.44 and 7.08 (2d, $J \approx 2.5$, H-C(7), H-C(9)). MS: 246 (100, M^+), 231 (4), 218 (54), 203 (7), 190 (36), 161 (7), 147 (12), 119 (9), 115 (7), 104 (5), 91 (7), 89 (7), 69 (9), 63 (8), 55 (6). Anal. calc. for C₁₄H₁₄O₄ (246.26): C 68.28, H 5.73; found: C 68.38, H 5.82.

5.3. Reaction with the 2-Bromosubstituted Monoacetal **30**. A solution of **30** (343 mg, 1.47 mmol), **27** (330 mg, 1.79 mmol), and (Bu)₄N⁺F⁻·3H₂O (100 mg, 0.3 mmol) in dry THF (8 ml) was boiled under reflux for 7 d (Ar). The residue obtained by evaporation was treated with 10 ml of a mixture of CH₃OH (10 ml) ClCH₂CH₂Cl (20 ml) and CH₃SO₃H (0.1 ml) [7] for 24 h under reflux. Evaporation and chromatography (50 g of silica gel, cyclohexane/CH₂Cl₂/AcOEt 1:2:2) gave 122 mg (41%) of 2-bromo-4-methoxyphenol and 71 mg (16%) of 6-bromo-8-methoxy-1,2,3,4-tetrahydrodibenzofuran-1-one (**34**): m.p. 154–155° (Et₂O/hexane). IR (CHCl₃): 3090w, 3030w, 3000m, 2960m, 2930m, 2870w, 2855w, 2840m, 1670s, 1612m, 1590s, 1481s, 1454m, 1443m, 1426s, 1393m, 1358w, 1342w, 1330w, 1316w, 1280m, 1262w, 1178s, 1169s, 1145w, 1125s, 1056m, 1033s, 1019s, 1002s, 904w, 885w, 965m, 850m, 644w. ¹N-NMR (80 MHz, CDCl₃): 2.05–2.45 (m, 2H-C(3)); 2.45–2.75 and 2.9–3.2 (2m, 2H-C(2), 2H-C(4)); 3.85 (s, CH₃O-C(8)); 7.06 and 7.45 (2d, $J \approx 3$, H-C(7), H-C(9)). MS (di.): 296 (99, M^+), 294 (100, M^+), 281 (3), 279 (3), 268 (78), 266 (78), 252 (7), 250 (19), 240 (46), 238 (46), 224 (6), 222 (17), 212 (5), 210 (5), 200 (8), 194 (11), 169 (6), 167 (6), 159 (24), 116 (12), 115 (11), 88 (10), 63 (12).

6. Reaction of 1-(Trimethylsilyloxy)cyclohexene (35) with Monoacetal 4. – 6.1. Reaction Catalyzed by CF_3SO_3TMS . A solution of 4 (597 mg, 3.87 mmol) and 35 (0.97 ml, *ca*. 5 mmol) in dry CH_2Cl_2 (4 ml) was added to a cooled (-78°) solution of CF_3SO_3TMS (2.8 ml, *ca*. 15 mmol) and Et_3N (2.3 ml, *ca*. 16 mmol) in CH_2Cl_2 (13 ml). After stirring for 19 h at r.t., the reaction was quenched by the addition to 1M aq. HCl and worked up with CH_2Cl_2 . The residue of the org. phases was treated over night with Ac_2O /pyridine (*ca*. 2 ml of each). Evaporation and chromatography (100 g of silica gel, hexane/acetone 3:1) gave 270 mg (42%) of 4-me-thoxyphenyl acetate (37) and 99 mg (9%) of 4-methoxy-3-(2'-oxocyclohexyl)phenyl acetate (36): IR (CHCl_3): 3030w, 2995m, 2935s, 2860m, 2830m, 1750s, 1708s, 1610w, 1491s, 1460m, 1445m, 1439w, 1420m, 1367s, 1335w, 1317m, 1298w, 1275w, 1190s, 1174s, 1153m, 1124m, 1063m, 1031s, 1011m, 950m, 899m, 886m, 886w, 835w. ¹H-NMR (80 MHz, CDCl_3): 1.5-2.4 (m, 6H); 2.25 (s, CH_3COO-C(1)); 2.3-2.7 (m, 2H-C(3')); 3.75 (s, CH_3O-C(4)); 3.6-4.15 (m, H-C(1')); 6.81 (d, J = 9, H-C(5)); 6.85 (d, J ≈ 3, H-C(2)); 6.98 (dd, J = 9 and ≈ 3, H-C(6)). MS: 262 (8, M^+), 220 (100), 192 (21), 176 (28), 163 (24), 161 (13), 150 (10), 137 (15), 135 (8), 123 (10), 107 (18), 91 (12), 77 (12), 43 (21).

6.2. Reaction Catalyzed by F^- . A solution of monoacetal 4 (284 mg, 1.843 mmol) and 35 (313 mg, 1.82 mmol) in 2 ml of dry THF was added to a cooled (-78°) solution of (Bu)₄N⁺F⁻·3H₂O (51 mg, 0.19 mmol) in THF (10 ml). After stirring for 6 h at -78° and for 14 h at r.t., the mixture was worked up with CH₂Cl₂. Chromatography (20 g of silica gel, hexane/acetone 3:1) gave 87 mg (18% based on 4) of 13,13-dimethoxy-8-oxatricyclo[7.3.1.0^{2.7}]tridec-2(7)-en-11-one (38): m.p. 102-103° (Et₂O/hexane). IR (CCl₄): 2990m, 2960s, 2938s, 2860m, 2835s, 1718s, 1692s, 1450m, 1442m, 1414w, 1395m, 1363m, 1322w, 1284m, 1218m, 1196s, 1180w, 1173w, 1149s, 1132w, 1109s, 1065s, 1046m, 1013m, 997m, 964m, 915w, 903w, 880m, 690m. ¹H-NMR (100 MHz, CDCl₃): 1.3–1.8 (m, 2H–C(4), 2H–C(5)); 1.75–2.2 (m, 2H–C(3), 2H–C(6)); 2.2–2.85 (m, H–C(1), 2H–C(10), 2H–C(12)); 3.30 and 3.31 (2s, 2CH₃O–C(13)); 4.41 (dt, $J \approx 4.5$ and 3, H–C(9)). ¹³C-NMR (75 MHz, CDCl₃): 22.7 and 22.8 (C(4), C(5)); 26.5 and 26.8 (C(3), C(6)); 37.8 (C(1)); 41.2 and 45.3 (C(10), C(12)); 48.2 and 48.8

(2 CH₃O); 70.6 (C(9)); 97.0 and 105.0 (C(2), C(13)); 144.3 (C(7)); 206.9 (C(11)). MS (di.): 252 (48, M^+), 221 (7), 220 (7), 181 (9), 165 (14), 135 (17), 101 (100), 91 (10), 79 (8), 77 (7), 67 (8), 55 (32), 41 (13). Anal. calc. for C₁₄H₂₀O₄ (252.31): C 66.65, H 7.99, O 25.36; found: C 66.65, H 7.93, O 25.22.

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